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and Solutions.

suspension. colloid.

Brownian motion.

Tyndall effect. a type of

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heterogeneous mixture  
whose particles settle  
out over time. a

heterogeneous mixture  
of intermediate-sized  
particles (between the  
erratic, random,  
movements of colloid  
particles that result in

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bridgetwalswick.

Prentice Hall. Terms in

this set (82) Gases are

easily \_\_\_\_\_, or

squeezed into a smaller

volume because of the

\_\_\_\_\_ between particles

in a gas. compressed;

space.

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MANUAL 32. Explain the similarities and differences between a 1 M solution of NaOH and a 1 m solution of NaOH. Both solutions contain NaOH (solute) dissolved in water (solvent). The 1 m solution contains 1 mole

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of NaOH per kilogram  
of water; the 1M  
solution 3 \_ 5 \_ \_ \_ 1 \_  
\_ \_ \_ \_ ( \_

*Mixtures and Solutions*  
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University. College of  
Staten Island CUNY.

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soluble. insoluble.  
miscible. immiscible.  
describes a substance  
that can be dissolved in*



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Answers

describes a substance  
that cannot be dissolved  
in a given solv....

describes two liquids  
that are soluble in each  
other.

*solutions quiz chapter  
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Organic Chemistry  
Chapter 14. Alkoxy  
Group.

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Chapter 14  
Answers  
Alkoxymercuration. ?

cleavage. Autoxidation.

A substituent consisting

of an alkyl group

bonded through an o....

Addition of mercury and

an alkoxy group to a

double bond, usua.... The

breaking of a bond

between the first and

second carbon ato....

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CHAPTER 14:  
Chemical Kinetics  
SOLUTIONS TO  
PROBLEMS

CHEM:1120 (Fall 2014)

1 1 14.1 Factors That  
Affect Reaction Rates

14.2 Reaction Rates

14.3 Concentration and  
Rate Laws 14.4 Change  
of Concentration with  
Time 14.5 Temperature  
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Mechanisms 14.7

Catalysis Chapter 14

Chemical Kinetics Ch.

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QUESTIONS 14.1.

Glucose or sucrose are  
soluble in water but  
cyclohexane and  
benzene (simple six

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Chapter 4  
Answers  
membered ring  
compounds) are  
insoluble in water

Explain. Ans:

The solubility of a solute in a given solvent follows the rule 'Like dissolves like'. Glucose contains five and sucrose contains eight -OH groups.

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## *Chapter 14* 14

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Biomolecules. Question

1. Carbohydrates can be divided into three major classes

monosaccharides,

oligosaccharides and

polysaccharides.

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access to the target*  
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$\ln\left[\frac{A}{[A]_0} - kt\right] = \frac{T}{2}$   
 $= \ln 2/k$ .  $k = Ae^{-E_a/RT}$ . 1.

The reaction of  
hydrogen ...

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Page 630 Problem 14-4  
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that the student invested countless hours, only to produce abysmal results. Often, inefficient study habits are to blame. The important question is: why do so many students have difficulty preparing themselves for organic chemistry exams? There are certainly several factors at play here, but perhaps the most dominant

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factor is a fundamental disconnect between what students learn and the tasks expected of them. To address the disconnect in organic chemistry instruction, David Klein has developed a textbook that utilizes a skills-based approach to instruction. The textbook includes all of the concepts typically

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covered in an organic chemistry textbook, but special emphasis is placed on skills development to support these concepts. This emphasis upon skills development will provide students with a greater opportunity to develop proficiency in the key skills necessary to succeed in organic chemistry. As an

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example, resonance structures are used repeatedly throughout the course, and students must become masters of resonance structures early in the course. Therefore, a significant portion of chapter 1 is devoted to drawing resonance structures. Two chapters (6 and 12) are devoted almost entirely to skill



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development. Chapter 6 emphasizes skills that are necessary for drawing mechanisms, while chapter 12 prepares the student for proposing syntheses. In addition, each chapter contains numerous Skillbuilders, each of which is designed to foster a specific skill. Each skillbuilder contains

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three parts: 1. Learn the Skill: a solved problem that demonstrates a particular skill; 2. Practice the Skill: numerous problems (similar to the solved problem) that give the students an opportunity to practice and master the skill; 3. Apply the Skill: one or two more-challenging problems in which the student must

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apply the skill in a slightly different environment. These problems include conceptual, cumulative, and applied problems that encourage students to think out of the box. Sometimes problems that foreshadow concepts introduced in later chapters are also included. All SkillBuilders are

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